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Jet Fuel from H₂O, CO₂ and Solar Energy

The entire production chain for renewable kerosene obtained directly from sunlight, H₂O, and CO₂ has been experimentally demonstrated. The key component of the production process is a high-temperature solar reactor containing a reticulated porous ceramic (RPC) structure made of ceria, which enables the splitting of H₂O and CO₂ via a 2-step thermochemical redox cycle. In the 1st reduction step, ceria is endo-thermally reduced using concentrated solar radiation as the energy source of process heat. In the 2nd oxidation step, nonstoichiometric ceria reacts with H₂O and CO₂ to form H₂ and CO – syngas – which is finally converted into kerosene by the Fischer-Tropsch process. The RPC featured dual-scale porosity for enhanced heat and mass transfer: mmsize pores for volumetric radiation absorption during the reduction step and μm-size pores within its struts for fast kinetics during the oxidation step. We report on the engineering design of the 4 kW solar reactor and the experimental demonstration of over 290 consecutive redox cycles for producing high-quality syngas suitable for the processing of liquid hydrocarbon fuels.

Reference: Marxer D., Furler P., Scheffe J., Geerlings H., Falter C., Batteiger V., Sizmann A., Steinfeld A., "Demonstration of the entire production chain to renewable kerosene via solar thermochemical splitting of H₂O and CO₂", Energy & Fuels 29, pp. 3241-3250, 2015.

Thursday, July 7, 2016

11:00 am – 12:00 Noon

Wrigley Hall 481 (ASU Tempe)

Light refreshments will be provided

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